REMARKS

Original Claim 1 is drawn to a process that includes contacting a 2-alkylhydroxyaromatic compound with "a BF₃ source which is capable of complex formation with the 2-alkylhydroxy compound".

The Office has taken the position that <u>Lange I</u> (U.S. 6,875,897) and <u>Lange II</u> (U.S. 6,914,163) describe the BF₃ source recited in the present claims. Applicants traverse the rejection for the reason that the Office provided no evidence that the catalysts of the <u>Lange</u> references are capable of forming a complex with the 2-alkylhydroxy aromatic compound recited in Claim 1.

Lange I makes it absolutely clear that the alkylation catalyst must be used in the presence of an ether co-catalyst. The ethers suitable for co-catalyst use in the Lange I process are described at column 3, lines 12-24. Applicants submit that the combination of BF₃ and one of the ether co-catalysts described in the Lange I reference forms a complex that is not capable of further complexing the 2-alkylhydroxy compound recited in present Claim 1. Because BF₃ is a strong Lewis acid and the ether co-catalysts of the Lange I reference are strong Lewis bases, the resulting complex is thermodynamically stable. The ether co-catalyst is thus not susceptible to substitution with the sterically hindered 2-alkylhydroxy compound.

Applicants submit that the rejection of the claims in view of <u>Lange I</u> should be withdrawn at least for the reason that the Office failed to demonstrate that the <u>Lange I</u> catalyst is capable of forming a complex with a 2-alkylhydroxy aromatic compound.

Applicants draw the Office's attention to new dependent Claim 23. New Claim 23 further defines the BF₃ source recited in Claim 1. Claim 23 excludes the catalyst/ether co-catalyst materials described in the <u>Lange I</u> reference. The subject matter of Claim 23 is thus further patentable over <u>Lange I</u>.

The rejection of the claims as anticipated by <u>Lange II</u> is likewise not supportable.

<u>Lange II</u> discloses the inclusion of complexes of BF₃ in a manner that would exclude the formation of a complex of the catalyst with the 2-alkylhydroxy aromatic compound of Claim 1 (see column 3, line 66 - column 4, line 24 - "<u>Ethers</u> suitable as cocatalysts..."). Applicants thus further respectfully request withdrawal of the rejection in view of the <u>Lange II</u> reference.

Not only does the cited art fail to disclose the presently-claimed invention, Applicants have demonstrated the significantly superior performance of the presently claimed process in comparison to processes which use a complex catalyst including an ether. Each of Examples 1-2 of the specification describe processes that are carried out with a catalyst that is a complex of BF₃ with a compound such as phenol (Comparative Example 1) or diethyl ether—"Alkylation using BF₃-diethyl ether complex as catalyst" (Comparative Example 2). Inventive Examples 4-7, on the other hand, include a BF₃ source that is capable of complex formation with a 2-alkylhydroxyaromatic compound.

When the resulting 2-alkylpolyisobutenylphenol is used in an additive for turbine fuel, a substantially reduced amount of particles are generated by thermal stress (see page 23, lines 19-20 of the specification). Applicants have thus not only demonstrated that the presently claimed invention is not anticipated over the <u>Lange</u> references but have further rebutted any as yet unstated rejection of the claims as obvious over the <u>Lange</u> references.

The rejection of claim 1 under 35 U.S.C. § 112, 2nd ¶, has been obviated by amendment. Accordingly, the rejection is no longer tenable and should be withdrawn.

The rejection of claims 19-20 and 22 under 35 U.S.C. § 101 has been obviated by amendment. Accordingly, the rejection is no longer tenable and should be withdrawn..

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For the reasons stated above, Applicants respectfully request withdrawal of the rejection and the allowance of all now-pending claims.

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